

Reminders 04-23-08:

- Exam 3 Average 65.0%**
- Exam 4 Monday May 5**
- Final Exam Wednesday May 7**

Outline:

- Line Spectra and the Bohr Model of Atom**
- Franck-Hertz Experiment**
- DeBroglie Waves**

- What should the energy of photons be so that the maximum change in wavelength due to Compton scattering by electrons is 1%?
 - Using these photons, what is $\Delta\lambda$ in angstroms for Compton scattered photons at 60° ?
 - What is the energy of the recoil electrons in this case?

$$\Delta\lambda = \lambda' - \lambda_0 = .0243\text{\AA} (1 - \cos\theta)$$

$$= \frac{h}{m_e c} (1 - \cos\theta)$$

$$1\text{\AA} = 10^{-10}\text{ m}$$

$$\frac{\Delta\lambda}{\lambda} = .01 = \frac{.0243(1 - \cos\theta)}{\lambda}$$

$$\theta = \pi$$

$$.01 = \frac{.0243(2)}{\lambda}$$

$$\lambda = \frac{.0486}{.01} = 4.86\text{\AA}$$

$$E_r = hf = \frac{hc}{\lambda} = \frac{(6.626 \times 10^{-34})(3 \times 10^8)}{4.86 \times 10^{-10}\text{ m}} \frac{1\text{ eV}}{1.6 \times 10^{-19}\text{ J}}$$

$$E = 2550\text{ eV}$$

Recoil energy not relativistic

$$\vec{p}_{1r} = \vec{p}_{2r} + \vec{p}_e$$

$$\vec{p}_c = \vec{p}_{1r} - \vec{p}_{2r}$$

$$p_c^2 = p_{1r}^2 + p_{2r}^2 - 2p_{1r}p_{2r}\cos\theta$$

$$p_c^2 = (p_{1r} + p_{2r})^2 = \left(\frac{h}{\lambda_1} + \frac{h}{\lambda_2}\right)^2$$

$$\text{If } \lambda_1 = 4.86 \text{ \AA}$$

$$\lambda_2 = ?$$

$$\lambda_2 = 4.86 - (0.1)(4.86)$$

$$\lambda_2 = 4.81 \text{ \AA}$$

$$p_c^2 = 1.86 \times 10^{-48} \text{ kg m s}$$

$$E = \frac{p^2}{2m} = \frac{1.86 \times 10^{-48}}{2(9.11 \times 10^{-31})} \cdot \frac{1}{1.6 \times 10^{-19}}$$

$$= 6.39 \text{ eV}$$

$$h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}$$

$$h = \frac{6.626 \times 10^{-34}}{1.6 \times 10^{-19}} = 4.14 \times 10^{-15} \text{ eV}\cdot\text{s}$$

$$\text{If } \lambda \text{ in nm } hc = 1240$$

$$\text{If } \lambda \text{ in \AA } hc = 12400$$

$$E = K + U = \underline{\hspace{2cm}} - \underline{\hspace{2cm}} \quad (1)$$

$$\Sigma F = ma \rightarrow \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \quad (2)$$

Use (2) to get $K = \underline{\hspace{2cm}}$

Use that result substitute into (1) to get

$$E = \underline{\hspace{2cm}} \quad (3)$$

Solve (2) for $v^2 = \underline{\hspace{2cm}}$

Solve the quantized L for $v^2 = \underline{\hspace{2cm}}$

Equate and solve for $r_n = \underline{\hspace{2cm}}$